Stereoselective Synthesis of Tetrahydropyran *D*-Ring of Methyl Sartortuoate

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A stereoselective synthesis of functionalized tetrahydropyran *D*-ring of methyl sartortuoate (1) was achieved starting from geraniol in a high yield. Sharpless asymmetric kinetic resolution, asymmetric dihydroxylation as well as asymmetric epoxidation were applied as key steps to establish all the four stereocenters of the *D*-ring.

Keywords methyl sartortuoate, stereoselective synthesis, Sharpless asymmetric kinetic resolution, Sharpless asymmetric dihydroxylation, Sharpless asymmetric epoxidation

Introduction

Methyl sartortuoate (1) and methyl isosartortuoate (2), two representative members of the structurally novel class of biscembranoids, were isolated from the marine Sarcophyton tortuosum tixierduriant by Su et al.^{1,2} (Figure 1). The relative configurations of 1 and 2 were elucidated by extensive NMR studies and their absolute stereochemistry was supported by X-ray analysis.¹ A preliminary bioassay proved that both of them displayed inhibitory effects against mouse S180 and cytotoxic activities towards KB cells.¹ It was hypothesized that 1 and 2 were formed by a biosynthetic Diels-Alder reaction of two cembrenanes as the precursors. This hypothesis could be supported by the isolation of methyl sarcoate (4), a dienophile unit of methyl sarcophytoate (3), although none of such precursors of 1 and **2** has been isolated to date.² Recently, Nakata *et al.*³ reported an elegant total synthesis of methyl sarcophytoate by an intermolecular Diels-Alder reaction, as well as the asymmetric syntheses of both the diene unit and the dienophile unit of **3**. For a long term concern, we⁴ initiated the total synthesis of 1 and 2 in order to investigate the potential bioactivity and the intriguing structural features as well as the interesting biogenetic possibility of them. Previously, we reported the asymmetric synthesis of the dienophile unit of **1** and **2** and the diene init of 1. Herein, we wish to describe an efficient synthesis of the diene unit of 1, which is necessary for the completion of the total synthesis of **1**.

Results and discussion

Although we have reported the synthesis of 6 from geraniol,^{4d} its total yield was pretty low, which ham-



Figure 1 The structures of selected biscembranoids 1-3 and methyl sarcoate 4.

pered our synthetic game. As shown in Scheme 1, we envisaged that 1 could be formed from D-A reaction from dienophile unit and diene unit 5, which, in turn, was synthesized from the key intermediate tetrahydropyran *D*-ring 6. Based on the retrosynthesis, a Sharpless asymmetric kinetic resolution of racemic allylic alcohol 8 available from geraniol furnished optically active epoxide 9 in over 40% yield in 94% *ee* (HPLC),⁵ which was subsequently protected with TESCl to give ether 10. Then the conversion of epoxide of 10 to vicinal diol that we previously described using modified Marshall's method with a sulfone-stablized allylic anion provided

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the desired tertiary alcohol in a decent yield. However, the subsequent allylic oxidation with SeO₂ only gave allylic alohol in a very poor yield with numerous side products. Thus, many nuclephilic reagents such as LiCH₂CN,⁶ NaCH(COOEt)₂,⁷ allylic metallic reagent⁸ were screened and tested to open epoxide 9, but it was found that the former two methods failed to give the desired products. Satisfyingly, success was realized when we treated 10 with allylic magnesium chloride in anhydrous THF at 40 °C for 4 h leading to tertiary alcohol 11 in an almost quantitative yield with no trace of chloride addition side product. It was noteworthy that Cu(I) catalyst was unnecessary for this step. Subsequent protection of the tertiary hydroxyl group with MOMCl of 11 afforded ether 12. To extend the left chain, regioselective dihydroxylation at the terminal double bond in the presence of a trisubstituted double bond was performed in advance. On the basis of the electron-withdrawing effect of the allylic substituent, a preference for dihydroxylation at the remote double bond under Sharpless asymmetric dihydroxylation condition was anticipated when the allylic alchohol was protected with acetyl group. Accordingly, removal of the benzyl group with lithium naphthalenide gave allylic alcohol 13, which was subsequently derived to acetate 14. Indeed, the dihydroxylation of acetate 14 under the standard condition (*i.e.* t-BuOH/H₂O, $V : V=1 : 1, 0 \ ^{\circ}C, 24 \ h)^{3}$ in the presence of (DHQD)₂PYR suppressed oxidation of the tri-substituted double bond giving the 1,2-diol (15) as a major product. This volume ratio was further increased to over 10:1 when (DHQD)₂PHAL or (DHQ)₂PHAL was used, while a 3:1 mixture was achieved by direct dihydroxylation of 13 in the absence of chiral ligand. Treatment of diol with NaIO₄ in THF/H₂O gave rise to aldehyde, and subsequent coupling with $Ph_3P = C(CH_3)COOEt$ in toluene afforded conjugated ester (18) (trans : cis > 95 : 5). Saponification of acetyl group, silylation of primary hydroxyl group with TBDPSCl, and reduction with DIBAL-H provided allylic alcohol (17), which was then selectively desilylated with PPTs/MeOH at 0 °C affording diol (18). Finally, Sharpless asymmetric epoxidation of 18

smoothly produced the desired tetrahydropyran *D*-ring (7) whose relative configuration was determined by a 2D-NOESY technique (Scheme 2).^{5,9}

Conclusion

In summary, an efficient asymmetric synthesis of 14-membered diene unit, required for the total synthesis of methyl sartortuoate, has been accomplished from geraniol. The total yield was increased to over 12%, while that of our previous synthetic route was only 4.0%. The synthesis is highlighted by using Sharpless asymmetric kinetic resolution, dihydroxylation and epoxidation to establish all the four stereocenters of the diene unit in a relatively high yield, which paved a way to the completion of the total synthesis of methyl sartortuoate. Further application of $\bf{6}$ to the total synthesis of methyl sartortuoate by a Diels-Alder reaction is in progress and will be reported in due course.

Experimental

General methods Unless stated otherwise, all reactions were carried out in dried glassware under a dry argon atmosphere. All solvents were purified and dried prior to use. Flash chromatography was performed on silica gel H (400 mesh). Optical rotations were measured on a precision automated polarimeter. Infrared spectra were recorded on an FT-IR spectrometer. ¹H NMR and ¹³C NMR spectra were recorded on a 300 MHz spectrometer. Chemical shifts are reported as δ values relative to internal chloroform (δ 7.26 for ¹H and 77.0 for ¹³C NMR). Elemental analyses, ESI-MS and ESI-HRMS were carried out at Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences.

(*R,E*)-6-(Benzyloxy)-4-methyl-1-[(*S*)-2-methyloxiran-2-yl]hex-4-en-1-ol (8) To a solution of 7 (13.0 g, 50 mmol) in CH₂Cl₂ (250 mL) were added molecular sieves (4 Å, 2 g) under argon. The solution was cooled to -25 °C and to this solution was added *D*-(-) DIPT (1.4 mL, 0.55 mmol) and (*i*-PrO)₄Ti (1.5 mL, 0.52 mmol). After stirring at -25 °C for 30 min, a solution of

Scheme 2



(a) Ref. 4e; (b) D-(-) DIPT, (*i*-PrO)₄Ti, TBHP, 4 Å MS, -20 °C, 40%; (c) TESCI, NEt₃, DMAP, CH₂Cl₂, r.t., 98%; (d) allylic magnesium chloride, THF, 40 °C, 4 h, 98%; (e) MOMCI, DIPEA, KI, DMAP, CH₂Cl₂, reflux, 6 h, 98%; (f) Li-naphthalenide, THF, 0 °C, 80%; (g) Ac₂O, NEt₃, DMAP, CH₂Cl₂, r.t., 95%; (h) K₂OsO₂(OH)₄, (DHQ)₂PHAL, K₃Fe(CN)₆, *t*-BuOH-H₂O, 0 °C, 24 h; (i) NaIO₄, THF-H₂O, r.t., then Ph₃P = C(CH₃)COOEt, toluene, > 95:5 (*trans:cis*), 69% (for 3 steps); (j) K₂CO₃, EtOH, r.t.; then TBDPSCI, NEt₃, DMAP, CH₂Cl₂, r.t., 96%; (k) DIBAL-H, THF, -78 °C, 97%; (l) PPTS, MeOH, 0 °C, 24 h, 92%; (m) *D*-(-)DIPT, (*i*-PrO)₄Ti, TBHP, 4 Å, MS, -20 °C, 2 h, 84%.

freshly prepared TBHP in CH₂Cl₂ (5.5 mL, 37 mmol) was added. The reaction mixture was stirred at -25 °C for 6 h and quenched with a solution of FeSO₄/tartaric acid. After stirring the mixture for 30 min at 0 $^{\circ}$ C, the organic layer was separated and the aqueous layer was extracted with ether (100 mL \times 3). The combined organic layers were treated with 100 mL of 30% NaOH/brine at 0 °C for 30 min. The organic layer was separated and the aqueous layer was extracted with ether (50 mL \times 3). The combined organic layers were washed with brine, dried over Na2SO4 and concentrated under reduced pressure. The residue was purified by flash chromatography to afford 8 (5.7 g, 40%, 94% ee) as a colorless oil. ¹H NMR (CDCl₃, 300 MHz) δ : 1.34 (s, 3H), 1.67 (s, 3H), 1.45-1.80 (m, 2H), 2.11-2.30 (m, 3H), 2.61 (d, J=4.8 Hz, 1H), 2.90 (d, J=4.8 Hz, 1H), 3.64 (d, 1H), 4.03 (d, J=6.9 Hz, 2H), 4.51 (s, 2H), 5.44(t, J=6.6 Hz, 1H), 7.31-7.47 (m, 5H).

(*R,E*)-1-Benzyl-3-methyl-6-[(*S*)-2-methyloxiran-2yl]-6-(triethylsilyloxy)hex-2-en-1-ol (9) To a solution of 8 (20 g, 47 mmol) and DMAP (480 mg, 4.0 mmol) in CH₂Cl₂ (200 mL) were added NEt₃ (10 mL, 73 mmol) and TESCI (8.6 mL, 52 mmol) at 0 °C. The resulting mixture was stirred for 2 h at room temperature and quenched with saturated NH₄Cl and the organic layer was separated. The aqueous phase was extracted with Et₂O (100 mL \times 3). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (aq.) (50 mL \times 2), saturated NaHCO₃, brine, dried (MgSO₄), and concentrated in vacuo. The residue was purified by flash chromatography to afford 9 (27.7 g, 98%) as a colorless oil. $[\alpha]_{D}^{20}$ +1.4 (c 1.0, CHCl₃); ¹H NMR (CDCl₃, 300 MHz) δ : 0.58 (q, J=7.8 Hz, 6 H), 0.95 (t, J=7.8 Hz, 9H), 1.29 (s, 3H), 1.74 (s, 3H), 1.61–1.80 (m, 2H), 2.01-2.20 (m, 2H), 2.58 (d, J=4.8 Hz, 1H), 2.68 (d, J=4.8 Hz, 1H), 3.24-3.28 (m, 1H), 4.03 (d, J=6.6 Hz, 2H), 4.51 (s, 2H), 5.43 (t, J=6.6 Hz, 1H), 7.26-7.36 (m, 5H); IR (film) v: 2995, 2877, 1455, 1116, 1076, 740 cm^{-1} ; EI-MS (70 eV) m/z (%): 391 (M+H⁺, 2), 359 (5). Anal. calcd for C₂₃H₃₈O₃Si: C 71.02, H 10.00; found C 70.72, H 9.81.

(5S,6R,E)-11-(Benzyloxy)-6-(triethylsilyloxy)-5,9dimethylundeca-1,9-dien-5-ol (10) To a solution of 9 (27.0 g, 69.0 mmol) in anhydrous THF (200 mL) was added 100 mL of Grinard reagent (2.0 mol \bullet L⁻¹, 200 mmol) at 0 $^{\circ}$ C. The resulting mixture was warmed to 40 °C. After being stirred for additional 2 h, the reaction mixture was diluted with 200 mL of ethyl ether and quenched with 200 mL of 1 mol \bullet L⁻¹ NH₄Cl (aq.). The organic layer was separated and the aqueous phase was extracted with Et₂O (200 mL \times 3). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (50 mL \times 2), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo. The residue was purified by flash chromatography to afford 10 (29.3 g, 98%) as a colorless oil. $[\alpha]_{D}^{20}$ +2.0 (c 1.0, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ : 0.58 (q, J=7.5 Hz, 6H), 0.98 (t, J=7.5 Hz, 9H), 1.12 (s, 3H), 1.65 (s, 3H), 1.40-1.70 (m, 4H), 1.95-2.30 (m, 4H), 3.46-3.50 (m, 1H), 4.02 (d, J=6.9 Hz, 2H), 4.51 (s, 2H), 4.93-5.08 (m, 2H), 5.41 (d, J=6.9 Hz, 1H), 5.79-5.92 (m, 1H), 7.26-7.36 (m, 5H); ¹³C NMR (CDCl₃, 75 MHz) δ : 5.41, 6.99, 16.58, 23.56, 27.77, 31.08, 35.51, 36.85, 66.51, 72.10, 74.48, 79.87, 114.15, 120.73, 127.45, 127.75, 128.28, 138.41, 139.19, 140.40; IR (film) v: 3476, 2956, 2914, 2877, 1641, 1455, 1097, 1006, 738 cm⁻¹; ESI-MS *m/z*: 450.3 $(M+NH_4^+)$, 455.3 $(M + Na^+)$. Anal. calcd for C₂₆H₄₄O₃Si: C 72.17, H 10.25; found C 72.03, H 10.12.

(6R,7S,E)-1-Benzyl-6-(triethylsilyloxy)-7-(methoxymethoxy)-3,7-dimethylundeca-2,10-dien-1-ol (11) To a solution of 10 (55.5 g, 130 mmol) in anhydrous CH₂Cl₂ (250 mL) was added DIPEA (85 mL, 490 mmol), DMAP (420 mg, 3.5 mmol) and KI (800 mg, 5.0 mmol). The reaction mixture was cooled to 0 $^{\circ}$ C, to which MOMCl (22 mL, 300 mmol) was added dropwise. After being stirred for 20 min at this temperature, the reaction mixture was refluxed for 8 h and cooled to r.t. The resulting mixture was quenched with saturated NH₄Cl and the organic layer was separated. The aqueous phase was extracted with Et_2O (300 mL×3). The combined organic layers were washed with 1 mol•L⁻ HCl (aq.) (100 mL \times 2), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo. The residue was purified by flash chromatography to yield ether 11 (60.0 g, 98%) as a colorless oil. $[\alpha]_{\rm D}^{20}$ +2.2 (c 1.0, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ : 0.61 (q, J=7.8 Hz, 6H), 0.96 (t, J=7.8 Hz, 9H), 1.18 (s, 3H), 1.66 (s, 3H), 1.40-1.90 (m, 4H), 1.95-2.30 (m, 4H), 3.36 (s, 3H), 3.52-3.56 (m, 1H), 4.03 (d, J=6.6 Hz, 2H), 4.51(s, 2H), 4.64 (d, J=6.9 Hz, 1H, B of AB), 4.79 (d, J=6.9 Hz, 1H, A of AB), 4.92-5.05 (m, 2H), 5.42 (d, J=6.6 Hz, 1H), 5.80-5.89 (m, 1H), 7.26-7.36 (m, 5H); ¹³C NMR (CDCl₃, 75 MHz) δ: 5.54, 7.06, 16.53, 19.09, 27.55, 31.02, 35.58, 37.32, 55.49, 66.51, 71.98, 77.15, 80.46, 91.18, 114.05, 120.56, 127.44, 127.72, 128.25, 138.43, 138.86, 140.76; IR (film) v: 3066, 2955, 2878, 1455, 1145, 1113, 1033, 919, 737 cm⁻¹; ESI-MS m/z: 494.3 $(M+NH_4^+)$, 499.3 $(M+Na^+)$. Anal. calcd for C₂₈H₄₈O₄Si: C 70.54, H 10.15; found C 70.72, H 10.30.

(6R,7S,E)-6-(Triethylsilyloxy)-7-(methoxymethoxy)-3,7-dimethylundeca-2,10-dien-1-ol (12) To a solution of 11 (24.8 g, 52 mmol) in anhydrous THF (200 mL) was added freshly prepared lithium naphthalenide in THF at -25 °C. After the reaction mixture became blue for 30 min, it was guenched with saturated NH₄Cl. The organic layer was separated and the aqueous phase was extracted with Et_2O (200 mL×3). The combined organic layers were washed with 1 mol•L HCl (aq.), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo. The residue was purified by flash chromatography to give alcohol 12 (15.8 g, 80%) as a colorless oil. $[\alpha]_{D}^{20}$ +26.2 (c 0.50, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ: 0.63 (q, J=8.1 Hz, 6H), 0.98 (t, J=8.1 Hz, 9H), 1.18 (s, 3H), 1.77 (s, 3H), 1.40-1.95 (m, 4H), 1.95-2.30 (m, 4H), 3.37 (s, 3H), 3.53-3.56 (m, 1H), 4.16 (d, J=6.6 Hz, 2H), 4.65 (d, J=7.2 Hz, 1H, B of AB), 4.79 (d, J=7.2 Hz, 1H, A of AB), 4.93–5.06 (m, 2H), 5.43 (t, J=6.6 Hz, 1H), 5.77—5.89 (m, 1H); 13 C NMR (CDCl₃, 75 MHz) δ : 5.53, 7.03, 16.25, 18.99, 27.52, 31.06, 35.49, 37.25, 55.45, 59.07, 77.09, 80.51, 91.13, 114.06, 123.28, 138.81, 139.60; IR (film) v: 3352, 2956, 2879, 1642, 1458, 1113, 1033, 1010, 740 cm⁻¹; ESI-MS m/z: 404.2 $(M+NH_4^+)$, 409.2 $(M + Na^+)$. Anal. calcd for C₂₁H₄₂O₄Si: C 65.24, H 10.93; found C 65.33, H 10.90.

(6R,7S,E)-6-(Triethylsilyloxy)-7-(methoxymethoxy)-3,7-dimethylundeca-2,10-dienyl acetate (13)To a solution of **12** (38.8 g, 100 mmol) in CH₂Cl₂ (200 mL) were added DMAP (480 mg, 4.0 mmol), NEt₃ (25 mL, 180 mmol) and Ac₂O (10.5 mL, 110 mmol) at 0 $^{\circ}$ C. The resulting mixture was stirred for 2 h at room temperature, quenched with saturated NH₄Cl and the organic layer was separated. The aqueous phase was extracted with Et₂O (200 mL \times 3). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (aq.), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo. The residue was purified by flash chromatography to provide acetate 13 (40.9 g, 95%) as a colorless oil. $[\alpha]_{D}^{20}$ +2.1 (c 1.0, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ: 0.63 (q, J=7.8 Hz, 6H), 0.97 (t, J=7.8 Hz, 9H), 1.18 (s, 3H), 1.76 (s, 3H), 1.38–1.86 (m, 4H), 2.06 (s, 3H), 1.95–2.35 (m, 4H), 3.27 (s, 3H), 3.52–3.56 (m, 1H), 4.59 (d, J=7.2 Hz, 2H), 4.65 (d, J=7.2 Hz, 1H, B of AB), 4.79 (d, J=7.2 Hz, 1H, A of AB), 4.93-5.05 (m, 2H), 5.36 (t, J=7.2 Hz, 1H), 5.76–5.89 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) *b*: 5.48, 6.98, 16.39, 18.98, 20.90, 27.49, 30.88, 35.47, 37.15, 55.42, 61.21, 77.00, 80.40, 91.14, 114.01, 118.02, 138.79, 142.54, 170.91; IR (film) $v: 2956, 2879, 1743, 1234, 1031, 918, 736 \text{ cm}^{-1};$ ESI-MS m/z: 446.2 (M+NH₄⁺), 451.2 (M+Na⁺). Anal. calcd for C₂₃H₄₄O₅Si: C 64.44, H 10.35; found C 64.31, H 10.25.

(2E,6S,7R,10E)-Ethyl 12-acetoxy-6-(methoxymethoxy)-2,6,10-trimethyl-7-(triethylsilyloxy)dodeca-2,10-dienoate (15) To a solution of 13 (15.7 g, 36.7 mmol) in *tert*-butyl alcohol were added H₂O (180 mL), K₃Fe(CN)₆ (36.7 g, 110 mmol), KC₂O₃ (16.1 g, 110

mmol) and (DHQ)₂ PHAL (140 mg, 0.19 mmmol). The reaction mixture was cooled at 0 $^{\circ}$ C for 10 min, then K₂OsO₂(OH)₄ (30 mg, 0.08 mmol) was added. After being vigorously stirred for 24 h at this termperature, the reaction mixture was quenched with 20 g of Na₂SO₃ and extracted with EtOAc (150 mL \times 4). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (aq.), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo to crude diol. The residue was then dissolved in THF (100 mL) and water (100 mL) and treated with NaIO₄ (9.1 g, 43 mmol) at room temperature for 2 h. The reaction mixture was filtered, the biscuit was washed with ethyl ether (30 mL \times 3) and the organic layer was separated. The aqueous phase was extracted with Et₂O (200 mL \times 3). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (aq.), saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo to give crude aldehyde, which was treated with $Ph_3P = C(CH_3)COOEt$ (10.0 g, 27.6 mmol) in toluene (60 mL) at room temperature. The reaction mixture was stirred overnight and concentrated in vacuo. The resulting residue was purified by flash chromatography to provide ester 15 (9.8 g, 69 %) as a colorless oil. $[\alpha]_D^{20}$ +3.4 (c 1.0, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ : 0.63 (q, J=8.1 Hz, 6H), 0.98 (t, J=8.1 Hz, 9H), 1.20 (s, 3H), 3.54–3.57 (m, 1H), 1.33 (t, J=7.5 Hz, 3H), 1.72 (s, 3H), 1.85 (s, 3H), 1.40-1.90 (m, 4H), 2.06 (s, 3H), 1.93-2.40 (m, 4H), 3.38 (s, 3H), 4.20 (q, J=7.5 Hz, 2H), 4.59 (d, J=7.2 Hz, 2H), 4.65 (d, J=7.2 Hz, 1H, B of AB), 4.80 (d, J=7.2 Hz, 1H, A of AB), 5.37 (t, J=7.2 Hz, 1H), 6.74 (t, J=7.5 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ : 5.47, 6.99, 12.21, 14.16, 16.42, 19.08, 20.91, 22.71, 30.93, 34.90, 37.12, 55.46, 60.26, 61.19, 77.12, 80.35, 91.09, 118.08, 127.57, 142.11, 142.43, 168.06, 170.94; IR (film) v: 2957, 2879, 1741, 1711, 1459, 1367, 1279, 1233, 1100, 1032, 741 cm⁻¹; ESI-MS m/z: 532.4 (M+NH₄⁺), 537.3 (M + Na⁺). Anal. calcd for C₂₇H₅₀O₇Si: C 63.00, H 9.79; found C 63.18, H 9.86.

(2E,6S,7R,10E)-Ethyl 7-(triethylsilyloxy)-12-(tertbutyldiphenylsilyloxy)-6-(methoxymethoxy)-2,6,10trimethyldodeca-2,10-dienoate (16) To a solution of 15 (10.4 g, 20.2 mmol) in ethanol (30 mL) was added K₂CO₃ (6.0 g, 43.3 mmol) at room temperature. After being stirred overnight, the reaction mixture was filtered and concentrated. The resulting residue was dissolved in CH₂Cl₂ (100 mL). To this solution were added NEt₃ (10 mL, 73 mmol), DMAP (240 mg, 2 mmol) and TBDPSCI (6.2 mL, 23.5 mmol) at 0 °C. The reaction mixture was warmed to r.t. and stirred overnight. After quenching the mixture with saturated NH₄Cl, the organic layer was separated. The aqueous phase was extracted with Et₂O (200 mL \times 3). The combined organic layers were washed with 1 mol \cdot L⁻¹ HCl (aq.) (100 mL \times 2), saturated NaHCO₃, brine, dried (MgSO₄), and concentrated in vacuo. The residue was purified by flash chromatography to afford ether 16 (13.8 g, 96 %) as a colorless oil. $[\alpha]_{D}^{20}$ +1.8 (c 1.0, CDCl₃); ¹H NMR $(CDCl_3, 300 \text{ MHz}) \delta$: 0.63 (q, J=7.8 Hz, 6H), 0.97 (t, J=7.8 Hz, 9H), 1.05 (s, 9H), 1.19 (s, 3H), 1.30 (t, J= 7.2 Hz, 3H), 1.44 (s, 3H), 1.40—1.85 (m, 4H), 1.84 (s, 3H), 1.90—2.30 (m, 4H), 3.37 (s, 3H), 3.53—3.57 (m, 1H), 4.19 (q, J=7.2 Hz, 2H), 4.21 (d, J=6.0 Hz, 2H), 4.65 (d, J=7.5 Hz, 1H, B of AB), 4.79 (d, J=7.5 Hz, 1H, A of AB), 5.39 (t, J=6.0 Hz, 1H), 6.74 (t, J=7.2 Hz, 1H), 7.34—7.47 (m, 6H), 7.68—7.71 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) δ : 5.58, 7.08, 12.29, 14.23, 16.35, 19.08, 19.16, 22.79, 26.75, 31.11, 34.97, 37.11, 55.51, 60.30, 61.04, 77.17, 80.45, 91.13, 123.92, 127.50, 127.61, 129.43, 133.92, 135.51, 137.23, 142.24, 168.09; IR (film) v: 2958, 2878, 1712, 1473, 1429, 1280, 1113, 1038, 740 cm⁻¹; ESI-MS m/z: 728.4 (M+NH⁴₄), 733.4 (M+Na⁺). Anal. calcd for C₄₁H₆₆O₆Si₂: C 69.56, H 9.26; found C 69.25, H 9.35.

(2E,6S,7R,10E)-7-(Triethylsilyloxy)-12-(tert-butyldiphenylsilyloxy)-6-(methoxymethoxy)-2,6,10-trimethyldodeca-2,10-dien-1-ol (17) To a solution of 16 (13.8 g, 19.4 mmol) in anhydrous THF (200 mL) at -78 °C was slowly added 50 mL of DIBAL-H (1 $mol \cdot L^{-1}$, 50 mmol) over 20 min. The reaction mixture was stirred for 20 min and quenched with 5 mL of water. The reaction mixture was diluted with 1.2 L of ethyl ether and 50 g of MgSO₄. After being stirred for 2 h, the reaction mixture was filtered and concentrated under reduced pressure. The residue was purified by flash chromatography to yield alcohol (17) (12.6 g, 97%) as a colorless oil. $[\alpha]_{D}^{20}$ +3.4 (c 1.0, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ : 0.64 (s, 6H), 0.97 (s, 9H), 1.05 (s, 9H), 1.19 (s, 3H), 1.44 (s, 3H), 1.70 (s, 3H), 1.40-1.80 (m, 4H), 1.90-2.20 (m, 4H), 3.37 (s, 3H), 3.53-3.56 (m, 1H), 4.01 (d, J=1.8 Hz, 2H), 4.21 (d, J=6.3 Hz, 2H), 4.68 (d, J=7.2 Hz, 1H, B of AB), 4.78 (d, J=7.2 Hz, 1H, A of AB), 5.40-5.44 (m, 2H), 7.34-7.45 (m, 6 H), 7.68–7.71 (m, 4H); IR (film) v: 3422, 3072, 2957, 1461, 921, 823 cm⁻¹; ESI-MS m/z: 686.5 (M+NH₄⁺), 691.5 (M+Na⁺). Anal. calcd for $C_{39}H_{64}O_5Si_2$: C 70.01, H 9.64; found C 70.05, H 10.03.

(2E,6S,7R,10E)-12-(tert-Butyldiphenylsilyloxy)-6-(methoxymethoxy)-2,6,10-trimethyldodeca-2,10-diene-1,7-diol (18) To a solution of 17 (4.2 g, 20.2 mmol) in methanol (20 mL) was added PPTS (200 mg, 0.8 mmol) at 0 °C. After being stirred overnight, the reaction mixture was concentrated and dissolved in 100 mL of EtOAc. The solution was washed with saturated NaHCO₃, brine, dried (MgSO₄), and concentrated in vacuo. The residue was purified by flash chromatography to afford diol 18 (3.2 g, 92 %) as a colorless oil. $[\alpha]_{D}^{20}$ +11.4 (c 0.50, CDCl₃); ¹H NMR (CDCl₃, 300 MHz) δ: 1.06 (s, 9H), 1.20 (s, 3H), 1.48 (s, 3H), 1.40-1.59 (m, 2H), 1.62 (s, 3H), 1.72-1.87 (m, 2H), 1.97-2.32 (m, 4H), 3.40 (s, 3H), 3.40-3.50 (m, 1 H), 4.01 (s, 2H), 4.24 (d, J=6.3 Hz, 2H), 4.66 (d, J=7.2 Hz, 1H, B of AB), 4.79 (d, J=7.2 Hz, 1H, A of AB), 5.40-5.44 (m, 2H), 7.34—7.45 (m, 6H), 7.68—7.71 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) *δ*: 16.49, 19.09, 20.44, 21.02, 23.00, 25.10, 26.79, 30.14, 34.98, 55.54, 61.01, 67.60, 73.11, 73.93, 74.98, 78.48, 90.70, 124.20, 127.51, 129.45, 133.88, 135.52, 136.58; IR (film) v: 3422, 2858, 1429, 1112, 1031, 740 cm⁻¹; EI-MS (70 eV) m/z (%): 553 (M-1, 0.66), 515 (M-39, 0.67). Anal. calcd for $C_{33}H_{50}O_6Si: C$ 71.44, H 9.08; found C 71.58, H 9.14.

(2R,3S,6S)-2-[(E)-5-tert-Butyldiphenylsilyloxy-3methylpent-3-enyl]-6-[(S)-1,2-dihydroxylpropan-2yl]-3-(methoxymethoxy)-3-methyl-tetrahydro-2H-pyran (6) To a solution of 18 (9.03 g, 16.3 mmol) in CH₂Cl₂ (200 mL) were added molecular sieves (4 Å, 1.5 g) under argon. The solution was cooled to -25 °C and to this solution were added D-(-) DIPT (4.6 g, 20 mmol) and (i-PrO)₄Ti (5.0 g, 17.8 mmol). After stirring at -25 °C for 30 min, a solution of freshly prepared TBHP in CH₂Cl₂ (6.45 mL, 5.0 mol•L⁻¹, 32 mmol) was added. The reaction mixture was stirred at -20 °C for 4 h and warmed to -10 °C and kept for an additional 2 h. The reaction mixture was quenched with a solution of FeSO₄/tartaric acid. After being stirred for 1 h at 0 °C, the organic layer was separated and the aqueous layer was extracted with Et₂O (100 mL \times 3). The combined organic layers were treated with 50 mL of 30% NaOH/brine at 0 °C for 30 min. The organic layer was separated and the aqueous layer was extracted with ether (100 mL \times 3). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated under reduced pressure. The residue was purified by flash chromatography to afford 6 (7.8 g, 84%) as a colorless oil. $[\alpha]_D^{20}$ +25 (c 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz) δ: 1.05 (s, 9H), 1.16 (s, 6H), 1.36-1.44 (m, 1H), 1.47 (s, 3H), 1.52-2.03 (m, 6H), 2.10-2.22 (m, 1H), 3.30 (d, J=11.2 Hz, 1H), 3.40 (s, 3H), 3.48–3.53 (m, 1H), 3.76–3.79 (m, 1H), 3.85 (d, J=11.2 Hz, 1H), 4.23 (d, J=6.3 Hz, 2H), 4.77 (s, 2H), 5.40 (t, J=6.3 Hz, 1H), 7.35-7.45 (m, 6H), 7.68-7.71 (m, 4H); ^{13}C NMR (CDCl₃, 75 MHz) δ: 16.49, 19.09, 20.44, 21.02, 23.00, 25.10, 26.79, 30.14, 34.98, 55.54, 61.01, 67.60, 73.11, 73.93, 74.98, 78.48, 90.70, 124.20, 127.51, 129.45, 133.88, 135.52, 136.58; IR (film) v: 3450, 3072, 2957, 2858, 1670, 1472, 823 cm⁻¹; HRMS (ESI) calcd for $(C_{33}H_{50}O_6Si + Na)^+$ 593.3269, found 593.3279.

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